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M.Sc
MCYE204

2nd Semester Back Examination 2017-18

ANALYTICAL TECHNIQUES

BRANCH : M.Sc.(AC)

Time : 3 Hours

Max Marks : 70

Q.CODE : C607

Question No.1 which is compulsory and any five from the rest

The figures in the right hand margin indicate marks.

Answer all parts of a question at a place.

- Q1** Answer the following questions : (2 x 10)
- a) Write the selection rules for electronic transitions.
 - b) What is the effect of polar solvent on $n \rightarrow \pi^*$ transition?
 - c) What is Fermi resonance in IR spectroscopy?
 - d) What is the NOE_{max} observed for a proton decoupled ^{13}C spectrum?
 - e) How many cross peaks are observed in the COSY-spectrum of 2-nitro propane?
 - f) What is the difference between molecular ion peak and base peak in mass spectrometry?
 - g) What is the spin selection rule in EPR spectroscopy?
 - h) What is FRET?
 - i) What do you mean by static and dynamic TGA?
 - j) What do you mean by retention factor in chromatography?
- Q2** a) What types of electronic transitions are possible for each of the following compounds? (6)
(i) Dimethyl ether, (ii) Triethylamine, (iii) Acetaldehyde
- b) The UV spectrum of acetone shows absorption maxima at 166, 189 and 279 nm. What type of transition is responsible for each of these bands? (4)
- Q3** Explain the instrumentation of IR spectrometer with block diagram. (10)
- Q4** Describe the applications of ^1H and ^{13}C NMR spectroscopy. (10)
- Q5** a) What is Nitrogen rule? (5)
b) Explain with example the McLafferty rearrangement. (5)
- Q6** a) What are Kramer's rule and zero field splitting? (7)
b) How many EPR signals are observed in methyl free radical? (3)
- Q7** Write the basic principle and instrumentation of a CD spectrometer. (10)
- Q8** Write short notes on : (5 x 2)
- a) Gas Chromatography
 - b) High performance Liquid Chromatography (HPLC)