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Total number of pages : 03

B.Tech.
PCE4I104

4th Semester Regular / Back Examination 2017-18
FUEL & ENERGY TECHNOLOGY
BRANCH : CHEM
Time : 3 Hours
Max Marks : 100
Q.CODE : C890

Answer Part-A which is compulsory and any four from Part-B.
The figures in the right-hand margin indicate marks.
Assume suitable notations and any missing data wherever necessary.
Answer all parts of a question at a place.

Part – A (Answer all the questions)

- Q1. Answer the following questions : (2 x 10)
- (a) The ultimate analysis of coal gives
 - i. Carbon, hydrogen, and ash
 - ii. Volatile matter, moisture, ash, and fixed carbon
 - iii. Carbon, hydrogen, sulphur, and nitrogen
 - iv. Volatile matter, moisture, nitrogen, and fixed carbon
 - (b) Which of the following coal has the highest calorific value?
 - i. Peat
 - ii. Lignite
 - iii. Bituminous
 - iv. Anthracite
 - (c) Control of incomplete combustion loss would mean
 - i. Decreasing the amount of excess air
 - ii. Increasing the amount of excess air
 - iii. Increasing the fuel/air ratio
 - iv. To decrease moisture loss
 - (d) The primary purpose of visbreaking process is
 - i. To increase viscosity and pour point
 - ii. To reduce viscosity and pour point
 - iii. To increase viscosity and reduce pour point
 - iv. To reduce viscosity and increase pour point
 - (e) The gaseous fuel that produces more heat than other fuel gases is
 - i. Water gas
 - ii. Producer gas
 - iii. Carburetted water gas
 - iv. Semi-water gas
 - (f) Which of the following hydrocarbons are the most desirable in kerosene?
 - i. Paraffins
 - ii. Isoparaffins
 - iii. Naphthenes
 - iv. Aromatics

- (g) The monometallic catalyst used in the catalytic reforming of naphthas is
- Pt
 - Ni
 - Fe
 - V_2O_5
- (h) Producer gas is a mixture of
- $CO + H_2$
 - $CH_4 + H_2$
 - $CO + N_2$
 - $CO + CH_4$
- (i) Heavy water is used in nuclear reactors to
- Cool the reactor
 - Facilitate the release of neutrons
 - Control fission
 - Slow down the speed of neutrons
- (j) Which of the following is not a fission fuel?
- U - 233
 - U - 235
 - Pu - 239
 - U - 238

Q2. Answer the following questions : (2 x 10)

- What are the objectives of coal washing?
- Mention petrographic constituents of coal.
- What is power alcohol?
- What are the advantages of catalytic cracking over thermal cracking?
- Write the various reactions involved in reforming process.
- Mention different types of coal tar fuels and uses of CTF.
- Write the composition and uses of natural gas.
- Mention the factors affecting composition of coke oven gas.
- In what way wind energy can be utilized?
- Write the properties of thorium.

Part – B (Answer any four questions)

- Q3.**
- Write the characteristics of Bituminous and Anthracite coal. (4)
 - Discuss the steps to be taken to prevent the loss of coal. (3)
 - What is washability of coal? Explain briefly any coal cleaning process. (8)
- Q4.**
- Discuss about the properties of coke. (5)
 - Briefly discuss about the byproducts of coke ovens for the manufacture of metallurgical coke. (10)
- Q5.**
- Describe in details about crude distillation system with neat flow diagram. (10)
 - Explain Delayed coking process with a neat sketch. (5)
- Q6.**
- Describe in detail, the Lurgi gasification process with a neat diagram of the gasifier. Also discuss in detail the process variables. (10)
 - Give a brief description on Fisher-Tropsch process. (5)

- Q7.** Discuss the functions of different elements of a nuclear reactor. Write in brief about the fast breeder reactor. **(15)**
- Q8.** The analysis of the coal in boiler trail was C = 81%, H₂ = 4.5%, O₂ = 8%, and remainder is incombustible. The Orsat analysis of the dry flue gas was CO₂ = 8.3%, CO = 1.4%, O₂ = 10%, N₂ = 80.3 %. Determine:
i. The weight of air supplied per kg of coal.
ii. The percentage of excess air. **(15)**
- Q9. (a)** How geothermal energy is used to generate electricity. **(5)**
(b) What is water gas? Explain its manufacturing process by a diagram with the reactions involved. How it differs from the carburetted water gas? **(10)**