F	Regis	stration No :	
		umber of Pages: 01  3 <sup>rd</sup> Semester Back Examination 2018-19 PHYSICAL PHRMACEUTICS – I BRANCH: B.Pharma Time: 3 Hours Max Marks: 100 Q.CODE: E913 er Question No.1 (Part-1) which is compulsory, any EIGHT from Part-II and a from Part-III. The figures in the right hand margin indicate marks.	B.Pharm 15PH301 any TWO
Q1	a) b) c) d) e) f) g) h) i)	Part- I Short Answer Type Questions (Answer All-10) Define Solution Define Solubility Define pH Define Isotonic Solution Define Critical Solution Temperature What do you mean by Stability What do you ean by polymorphism Whatdo you mem by TPD Define Isotoinc solution How would you express solubility	(2 x 10)
Q2	-	Part- II  Focused-Short Answer Type Questions- (Answer Any Eight out of Twelve) Discuss Buffers and its role in formulation How would you prepare buffers? How would you calculate Expiry date of Products Discuss Chemical Kinetics Discuss Eutectic mixture and its importance in formulation Describe Phase Rule. What do you mean by Critical temperature and triple point Discuss critical solution temperature and Lower Consulate temperature Describe a method of determine solubility Discuss two component system taking phenol waer system. What is Complexation, state its different types. What do you mean by Buffer capacity and how would you determine it.	(6 x 8)
Q3		Part-III  Long Answer Type Questions (Answer Any Two out of Four)  Describe methods for preparation of isotonic solution.	(16)
Q4		Describe method for conducting stability test of formulated product	(16)
Q5		Describe Van'stoff Equation and Clausius Clapeyron eq.	(16)
Q6		Discuss different state of matters also liquid crystal.	(16)