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B.Pharma. BP104T

1st Semester Regular Examination 2017-18 PHARMACEUTICAL INORGANIC CHEMISTRY BRANCH: B.Pharma.

Time: 3 Hours Max Marks: 75 Q.CODE: B1019

Answer Question No.1 which is compulsory and any five from the rest.

The figures in the right hand margin indicate marks.

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Q1	i) a) b) c) d) e) f) g) h) i) j)	Answer the following questions: Define the term antidote with example. What is Cathartics? What are electrolytes? Give example? What is the difference between self life and half life of a drug? What are Expectorants used for? Why is lead acetate moistened cotton is used in the Limit test of Arsenic? What is a radioactive isotope and what are they used for? Define the term acidifiers. Give Examples. Define Haematinics with two examples. What is a Monograph of a drug?	(2 x 10)
	<u>ii)</u> a)	Answer the following questions Impurities in pharmaceutical preparation may be due to following sources: (a) Raw material (b) Manufacturing process (c) Chemical instability (d) All of the above	(5×1)
	b) c)	Inorganic antimicrobial agent can be divided into (a) Oxidation (b) halogenation (c) Protein precipitate (d) all of the above In Bronsted-Lowry concept acid is	
	d) e)	a) 2010 b) 2014 c) 2007 d) 2018	
Q2		Discuss in brief the various sources of impurities. Write down the principle involved in the limit test for Chloride.	(10)
Q3		Define buffer solution. Mention the role of buffers in pharmacy. Define the term Isotonicity. How do you measure the tonicity of the fluid?	(10)
Q4		Mention the major intra and extracellular electrolytes used in replacement therapy. What is the composition of ORS as per WHO?	(10)
Q5	a)	Define dentifrices. Write down the use of Calcium carbonate and Zinc eugenol cement.	(5)
•	b)	What do you mean the term desensitizing agents with example?	(5)
Q6	a)	Define and classify antacids with examples.	(5)
	b)	Mention the ideal properties of an antacid. Write a note on combinations of	(5)

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Q7 a) Write short note on emetics and antidote.
b) Define and classify astringents with examples. Mention their mechanisms.
Q8 a) What is meant by radioactivity? How do you measure the radioactivity of a compound?
b) Write a note on α, β and γ radiation. Mention the imporatance of Sodium iodide
(5)
(5)